

Magnetism by Interfacial Hybridization and *p*-type Doping of MoS₂ in Fe₄N/MoS₂ Superlattices: A First-Principles Study

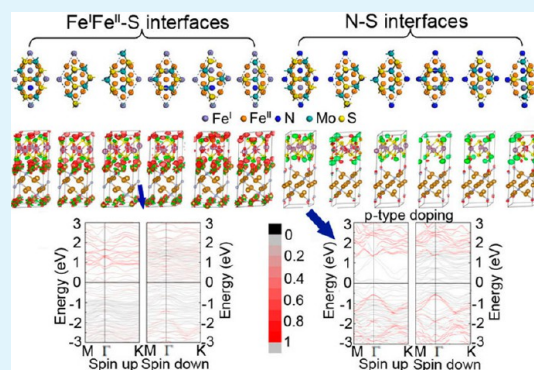
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ABSTRACT: Magnetic and electronic properties of Fe₄N(111)/MoS₂($\sqrt{3} \times \sqrt{3}$) superlattices are investigated by first-principles calculations, considering two models: (I) Fe^IFe^{II}-S and (II) N-S interfaces, each with six stacking configurations. In model I, strong interfacial hybridization between Fe^I/Fe^{II} and S results in magnetism of monolayer MoS₂, with a magnetic moment of 0.33 μ_B for Mo located on top of Fe^I. For model II, no magnetism is induced due to weak N-S interfacial bonding, and the semiconducting nature of monolayer MoS₂ is preserved. Charge transfer between MoS₂ and N results in *p*-type MoS₂ with Schottky barrier heights of 0.5–0.6 eV. Our results demonstrate that the interfacial geometry and hybridization can be used to tune the magnetism and doping in Fe₄N(111)/MoS₂($\sqrt{3} \times \sqrt{3}$) superlattices.

KEYWORDS: MoS₂, Fe₄N, electronic structure, magnetic properties



INTRODUCTION

Since the discovery of graphene, many studies have addressed two-dimensional materials.¹ Although the mobility of graphene can reach 1×10^5 cm²/(V s), lack of a band gap yields a small current on/off ratio that hampers its application in electronics devices.² Thus, two-dimensional materials with an intrinsic band gap are receiving more and more attention, such as monolayer MoS₂,^{3–9} due to its high thermal stability, no dangling bonds, a direct band gap of 1.8 eV, and giant spin splitting in the K valleys, hence being of interest for electronics, photonics, spintronics, and valleytronics.^{10,11} Recently a field-effect transistor based on monolayer MoS₂ with a high mobility of 200 cm²/(V s), current on/off ratio of 1×10^8 , and ultralow standby power dissipation has been demonstrated.¹²

Mn, Fe, Co, and Zn doping in monolayer MoS₂ are predicted to result in two-dimensional diluted magnetic semiconductors.¹³ However, as dilute doping is difficult experimentally, it is important to find other means to introduce magnetism in monolayer MoS₂. By the magnetic proximity effect,^{14,15} spin injection from a high spin-polarized ferromagnetic electrode seems to be a promising route. Yet, it can be expected that the interfacial interaction plays a crucial role, compare, for example, the different barrier heights for electron tunneling in MoS₂/Au and MoS₂/Ti contacts,¹⁶ *n*-doping of MoS₂ at Ir, Pd, and Ru interfaces,¹⁷ controllable Schottky barrier heights in MoS₂/Co transistors,¹⁸ and induced conductivity in MoS₂/Ti₂C.¹⁹ To achieve high-temperature spintronics devices, electrodes with high spin polarization and Curie temperature are necessary. In bulk Fe₄N these quantities amount to almost –100% and 761

K,^{20,21} respectively. In fact, magnetic tunnel junctions with Fe₄N electrodes are known for highly spin-polarized transport,^{22,23} whereas Fe₄N/MoS₂ heterostructures have not been addressed so far. In this paper, we therefore study the electronic structures and magnetism of Fe₄N(111)/MoS₂($\sqrt{3} \times \sqrt{3}$) superlattices by first-principles calculations and demonstrate that spin polarization is induced in monolayer MoS₂ in contact with Fe^I/Fe^{II} and *p*-type doping in contact with N.

CALCULATION DETAILS AND MODELS

Our first-principles calculations are based on density functional theory²⁴ and the projector augmented wave method²⁵ as implemented in Vienna Ab initio Simulation Package.²⁶ The Perdew–Burke–Ernzerhof²⁷ flavor of the spin-polarized generalized gradient approximation is used. The plane-wave cutoff is set to 500 eV and the Brillouin zone is sampled with $9 \times 9 \times 9$ and $9 \times 9 \times 1$ *k* meshes for bulk Fe₄N and the Fe₄N/MoS₂ superlattice, respectively. All structures are fully optimized until the force on each atom is less than 0.01 eV/Å and the total energy is converged to 1×10^{-5} eV.

Bulk Fe₄N has a cubic perovskite-type structure (*Pm3m*) with a lattice constant of $a = 3.795$ Å. Two types of Fe atoms (Fe^I and Fe^{II}) occupy the corner and face-centered sites, whereas N locates at the body-centered position. Fe^I is surrounded by 12 Fe^{II} atoms in a distance of $a/\sqrt{2}$ and Fe^{II}

Received: February 4, 2014

Accepted: February 27, 2014

Published: February 27, 2014

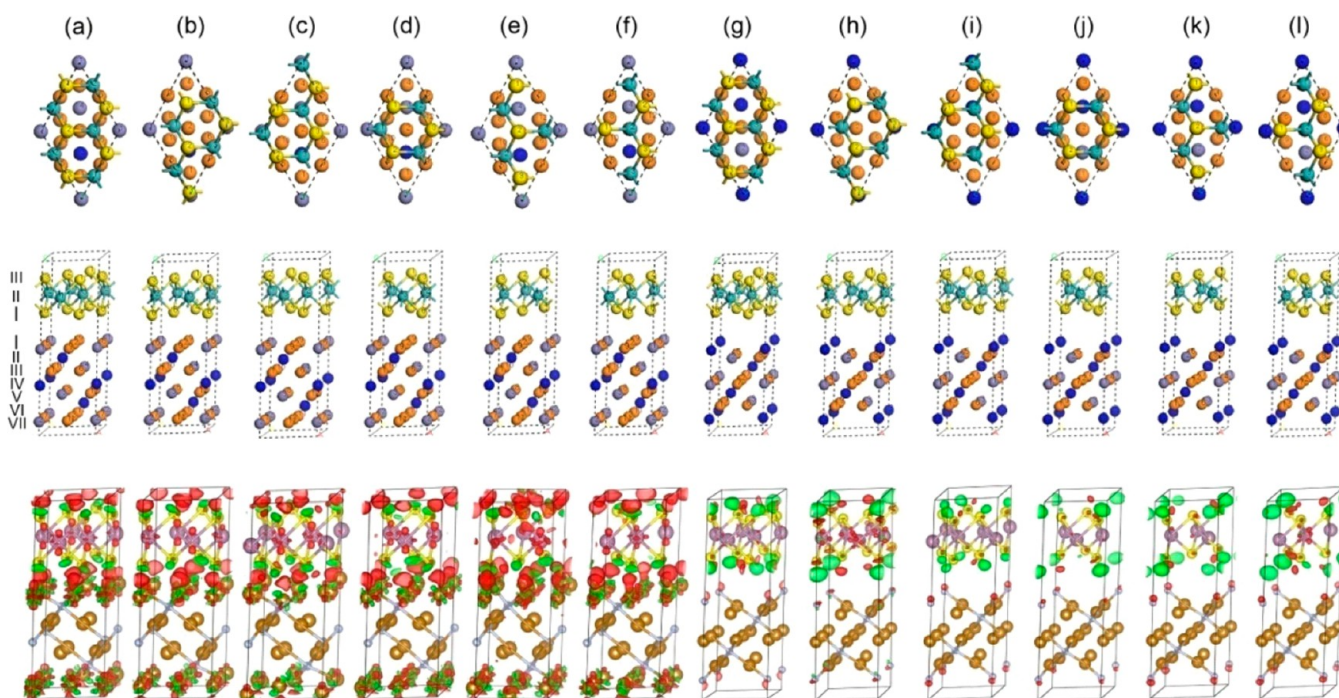


Figure 1. Top and side views as well as charge density difference. The red regions represent accumulation and the green regions depletion of electrons relative to the two isolated components. Light blue spheres stand for Fe^I, orange for Fe^{II}, navy blue for N, yellow for S, and cyan for Mo. (a–f) Fe^I/Fe^{II}–S interface (0.0056 e/Å³). (g–l) N–S interface (0.0002 e/Å³).

has two N atoms in a distance of $a/2$ as the nearest neighbors and 12 Fe^I atoms as the second nearest neighbors.²⁸ Bulk MoS₂ is a layered semiconductor with space group $P63mmc$ (D_{6h} point group), consisting of stacked S–Mo–S monolayers with weak van der Waals interaction. The indirect band gap of 1.2 eV for bulk MoS₂ transforms into a direct band gap of 1.8 eV for monolayer MoS₂ (D_{3h} point group). The S–Mo–S slab consists of hexagonally arranged Mo atoms sandwiched between two sheets of likewise hexagonally arranged S atoms. In the sheets, the atoms are covalently bonded with S in a trigonal prismatic environment.

Fe₄N/MoS₂ superlattices are built by attaching a monolayer of MoS₂ on top of one unit cell (7 monolayers) of Fe₄N(111). The surface unit cell of Fe₄N(111) has $p(1 \times 1)$ periodicity with a lattice constant of 5.366 Å, whereas monolayer MoS₂(0001) has $\sqrt{3} \times \sqrt{3}R30^\circ$ periodicity with a lattice constant of 5.489 Å, resulting in a small lattice mismatch. Thus, an average in-plane lattice constant of 5.428 Å is adopted. We chose the Fe^I/Fe^{II} and N surfaces as terminations. By the periodic boundary conditions, the supercells contain two interfaces. Model I is the Fe^I/Fe^{II}–S interface (see Figure 1a–f) and model II the N–S interface (see Figure 1g–l). To investigate how the electronic structure of monolayer MoS₂ depends on the interface geometry, we considered six stackings for each model, see Figure 1.

The stable pattern is determined by calculating the work of separation, i.e., the cohesive energy between Fe₄N and MoS₂, $W_{\text{sep}} = (E_{\text{Fe}_4\text{N}} + E_{\text{MoS}_2} - E_{\text{Fe}_4\text{N}/\text{MoS}_2})/2$, where $E_{\text{Fe}_4\text{N}/\text{MoS}_2}$ is the total energy of the superlattice, $E_{\text{Fe}_4\text{N}}$ and E_{MoS_2} represent the energies of the same supercell containing either the Fe₄N or MoS₂ parts (i.e., we keep the equilibrium structure obtained for the superlattice). The factor 2 accounts for the two interfaces present in the supercell. For illustrating the nature of the charge transfer at Fe₄N/MoS₂ interface, we calculate the charge

density difference by subtracting the charge densities of the isolated MoS₂ and Fe₄N parts from the charge density of the superlattice. The electronic structures of isolated MoS₂ and Fe₄N are calculated by freezing the atoms of the respective component at the superlattice positions. The work function (W) is defined by the smallest energy required to remove an electron deep inside bulk crystal. It is defined as $W = V(\infty) - E_{\text{F}}$, where $V(\infty)$ is the electrostatic potential in the middle of the vacuum region of our supercells and E_{F} is the Fermi energy.

RESULTS AND DISCUSSION

For bulk Fe₄N a lattice constant of 3.795 Å is obtained, which agrees well with the experimental value.²⁰ The calculated average magnetic moments of Fe^I, Fe^{II}, and N are 2.95, 2.35, and 0.03 μ_{B} , respectively, in accordance with previous theoretical and experimental results.^{29–33} Figure 2 presents the partial density of states (PDOS) for bulk Fe₄N and

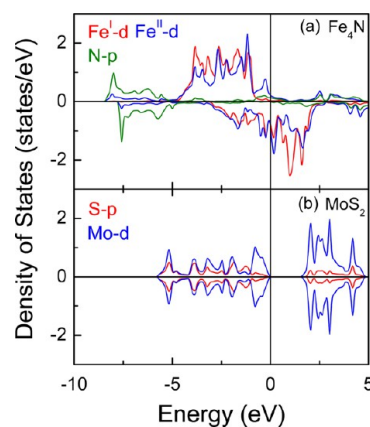


Figure 2. PDOS of (a) Fe₄N and (b) pristine monolayer MoS₂.

Table 1. Work of Separation W_{sep} (eV) and Optimized Bond Lengths d (Å) for Different Stackings, Compare Figure 1

| model | case | W_{sep} | $d_{\text{FeI-S}}$ | $d_{\text{FeII(2)-S}}$ | $d_{\text{FeII(5)-S}}$ | $d_{\text{FeII(9)-S}}$ | $d_{\text{N-S}}$ |
|---|------|------------------|--------------------|------------------------|------------------------|------------------------|------------------|
| Fe ₄ N/ MoS ₂ (I) | (a) | 1.74 | 2.81 | 2.30 | 2.30 | 2.30 | |
| | (b) | 1.65 | 2.33 | 2.47 | 2.89 | 2.20 | |
| | (c) | 2.05 | 2.53 | 2.28 | 2.28 | 2.28 | |
| | (d) | 1.68 | 2.37 | 2.86 | 2.36 | 2.21 | |
| | (e) | 2.04 | 2.29 | 2.32 | 2.38 | 2.26 | |
| | (f) | 1.65 | 2.33 | 2.89 | 2.19 | 2.45 | |
| Fe ₄ N/ MoS ₂ (II) | (g) | 0.01 | | | | | 3.75 |
| | (h) | 0.01 | | | | | 3.59 |
| | (i) | 0.02 | | | | | 3.91 |
| | (j) | 0.02 | | | | | 3.76 |
| | (k) | 0.02 | | | | | 3.76 |
| | (l) | 0.02 | | | | | 3.64 |

Table 2. Optimized Lattice Constants and Bond Angles

| case | a (Å) | b (Å) | c (Å) | α (deg) | β (deg) | γ (deg) |
|------|---------|---------|---------|----------------|---------------|----------------|
| (a) | 5.40 | 5.40 | 13.98 | 90.0 | 90.0 | 60.0 |
| (b) | 5.41 | 5.35 | 14.06 | 90.0 | 91.3 | 60.4 |
| (c) | 5.46 | 5.46 | 13.68 | 90.0 | 90.0 | 60.0 |
| (d) | 5.42 | 5.42 | 14.00 | 90.2 | 90.2 | 59.3 |
| (e) | 5.40 | 5.40 | 13.85 | 93.7 | 93.7 | 59.0 |
| (f) | 5.41 | 5.41 | 14.06 | 88.8 | 88.8 | 59.3 |
| (g) | 5.40 | 5.40 | 16.06 | 90.0 | 90.0 | 60.0 |
| (h) | 5.42 | 5.41 | 16.60 | 90.0 | 91.2 | 60.1 |
| (i) | 5.42 | 5.43 | 16.51 | 90.0 | 89.5 | 60.0 |
| (j) | 5.42 | 5.42 | 16.69 | 90.5 | 90.5 | 60.1 |
| (k) | 5.42 | 5.42 | 16.32 | 90.4 | 90.4 | 60.1 |
| (l) | 5.43 | 5.43 | 16.41 | 88.7 | 88.7 | 59.8 |

monolayer MoS₂. For Fe^I, Fe^{II}, and N (Figure 2a) a distinct spin splitting of Fe d states appears, which induces the ferromagnetism. The DOS of Fe₄N at E_{F} is mainly from the

spin-down channel. From -8.5 to -5 eV the N p states strongly hybridize with Fe^{II} d states to form covalent bonds, whereas in the same energy range the Fe^I d states show hardly any hybridization with N. Thus, Fe^{II} magnetic moment is smaller than that of Fe^I. The band structure of monolayer MoS₂ (1×1 unit cell) shows a direct band gap of 1.7 eV between the valence band (VB) maximum and conduction band (CB) minimum at K point of Brillouin zone, consistent with previous theoretical works^{13,34,35} and photoluminescence spectroscopy.³⁶ Figure 2b indicates strong hybridization between the Mo d and S p states in the VB and CB, where the VB maximum and CB minimum are governed by Mo d states.

In Table 1, the work of separation and optimized bond lengths are presented for different stackings. Monolayer MoS₂ strongly interacts with Fe^IFe^{II}-terminated surface, forming chemical bonds with significant configuration dependence. The stability order in model I is found to be $b < f < d < a < e < c$, see Figure 1. The energetically favorable case is c, i.e., Mo located on top of Fe^I. The large $W_{\text{sep}} = 2.05$ eV is explained by strong Fe^{II}-S bonds with a bond length of 2.28 Å. In contrast, for MoS₂ on the N-terminated surface the results indicate that the interaction is rather weak and fairly insensitive to the stacking. The optimized distances and bond angles point to strong distortions of MoS₂ and Fe₄N, where the S atoms either are pulled toward Fe₄N or pushed away, see Tables 1 and 2, which lowers the symmetry of Fe₄N.

We first focus on model I to explore the variations of the electronic structure induced by the interfacial coupling. As compared to the value of bulk Fe₄N, the magnetic moments decrease by 4.4, 4.8, 3.7, 3.1, 3.4, and 4.9% for cases a–f, respectively.

Usually, the appearance of Fe magnetic moments in Fe₄N is ascribed to symmetry breaking, as N is located at the body-centered site of the nonmagnetic face-centered cubic unit cell of Fe. So, the enhancement of the Fe magnetic moments in our

Table 3. Average Fe^I and Fe^{II} Magnetic Moments $\bar{m}(\mu_{\text{B}})$ for Each Fe₄N Layer in Model I; Magnetic Moments $m(\mu_{\text{B}})$ of Fe^I, Fe^{II}, and N at the Interfaces As Well as of S and Mo

| moment | atom | bulk | a | b | c | d | e | f | |
|-----------|-----------------------|-----------|-------|-------|-------|-------|-------|-------|-------|
| \bar{m} | Fe ^I | layer I | 2.95 | 2.83 | 2.74 | 2.82 | 2.80 | 2.68 | 2.76 |
| | | layer III | 2.95 | 2.90 | 2.87 | 2.83 | 2.88 | 2.88 | 2.87 |
| | | layer V | 2.95 | 2.90 | 2.87 | 2.84 | 2.89 | 2.87 | 2.87 |
| | | layer VII | 2.95 | 2.84 | 2.75 | 2.82 | 2.80 | 2.75 | 2.75 |
| | Fe ^{II} | layer I | 2.35 | 2.28 | 2.29 | 2.27 | 2.33 | 2.35 | 2.27 |
| | | layer III | 2.35 | 2.21 | 2.20 | 2.28 | 2.25 | 2.24 | 2.20 |
| | | layer V | 2.35 | 2.21 | 2.20 | 2.28 | 2.25 | 2.23 | 2.20 |
| m | Fe ^I (1,2) | layer VII | 2.35 | 2.28 | 2.29 | 2.27 | 2.32 | 2.35 | 2.31 |
| | | layer I | 2.95 | 2.83 | 2.74 | 2.82 | 2.80 | 2.68 | 2.76 |
| | | layer III | 2.35 | 2.27 | 2.53 | 2.27 | 2.58 | 2.46 | 2.59 |
| | | layer V | 2.35 | 2.24 | 2.59 | 2.27 | 2.43 | 2.27 | 1.67 |
| | | layer VII | 2.35 | 2.31 | 1.73 | 2.26 | 1.99 | 2.32 | 2.53 |
| | | N (2) | 0.03 | -0.03 | -0.03 | -0.02 | -0.02 | -0.01 | -0.03 |
| | | S(1) | 0 | 0.02 | 0.02 | 0.03 | 0.02 | 0.02 | 0.02 |
| | S(2) | | 0.03 | 0.01 | 0.03 | 0.01 | 0.03 | 0.02 | |
| | S(3) | | 0.03 | 0.01 | 0.03 | 0.02 | 0.02 | 0.01 | |
| | S(4) | | 0.03 | 0.01 | 0.03 | 0.01 | 0.01 | 0.01 | |
| | S(5) | | 0.02 | 0.02 | 0.03 | 0.02 | 0.03 | 0.01 | |
| | S(6) | | 0.03 | 0.02 | 0.03 | 0.02 | 0.02 | 0.02 | |
| | Mo(1) | | -0.03 | -0.09 | 0.22 | -0.04 | -0.04 | -0.09 | |
| Mo(2) | | 0.03 | 0.07 | -0.03 | 0.00 | -0.04 | 0.07 | | |
| Mo(3) | | -0.02 | -0.09 | -0.03 | 0.01 | -0.16 | -0.09 | | |

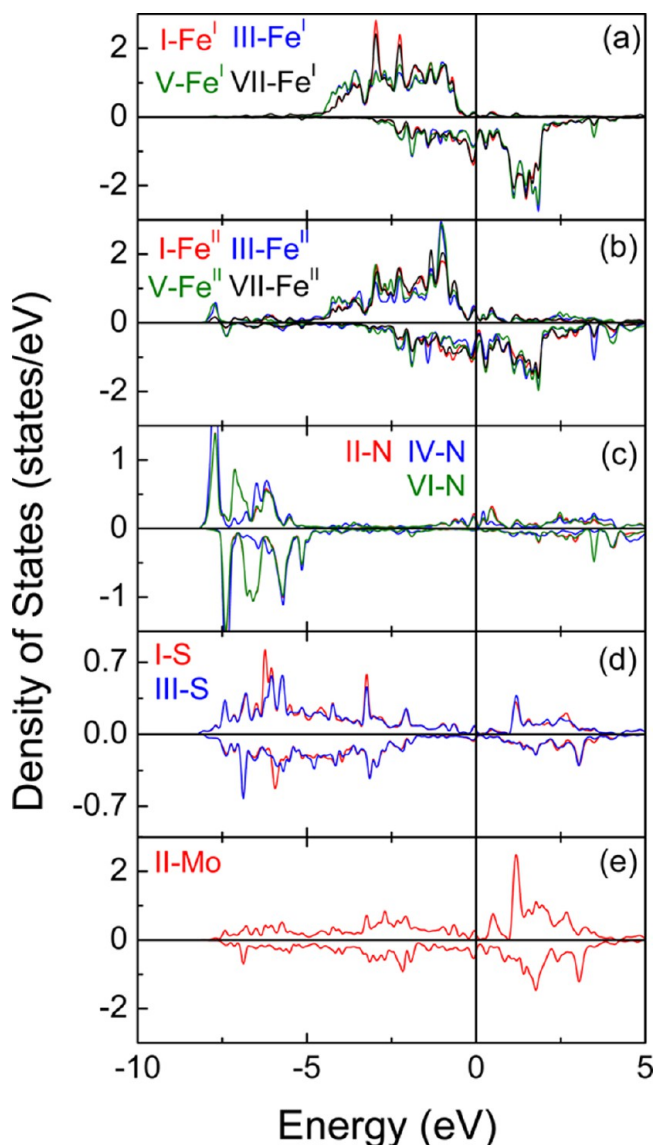


Figure 3. PDOS of the fully relaxed case a of the $\text{Fe}_4\text{N}/\text{MoS}_2$ superlattice: (a) Fe^{I} , (b) Fe^{II} , (c) N, (d) S, and (e) Mo. The Fermi energy is indicated by vertical lines and is set to 0 eV.

case should also be related to the symmetry. Since they mainly originate from unpaired d electrons and the enhanced hybridization between Fe and other atoms reduces the amount of unpaired d electrons, it is the hybridization that determines the spin polarization.³⁷ The average Fe^{I} and Fe^{II} magnetic moments for each Fe_4N layer in model I are shown in Table 3.

We take case a as an example to illustrate the similarities among the different stackings, see the PDOSs in Figure 3. The label I– Fe^{II} , for example, refers to layer I, as defined in Figure 1, and Fe^{II} denotes the Fe atoms at face-centered sites (distinguished by numbers in round brackets). Reduced Fe^{I} magnetic moments in layers I and VII are attributed to hybridization with S in layers I, III and enhanced hybridization with N in layers II, VI, respectively. Fe^{I} magnetic moments in layers III and V decrease due to enhanced hybridization with N in layers II, IV and IV, VI, respectively. As Fe^{I} in layers I and VII strongly hybridize with S, the Fe^{I} magnetic moments are slightly smaller than in layers III and V. The average Fe^{II} magnetic moment for each layer is smaller than the bulk value

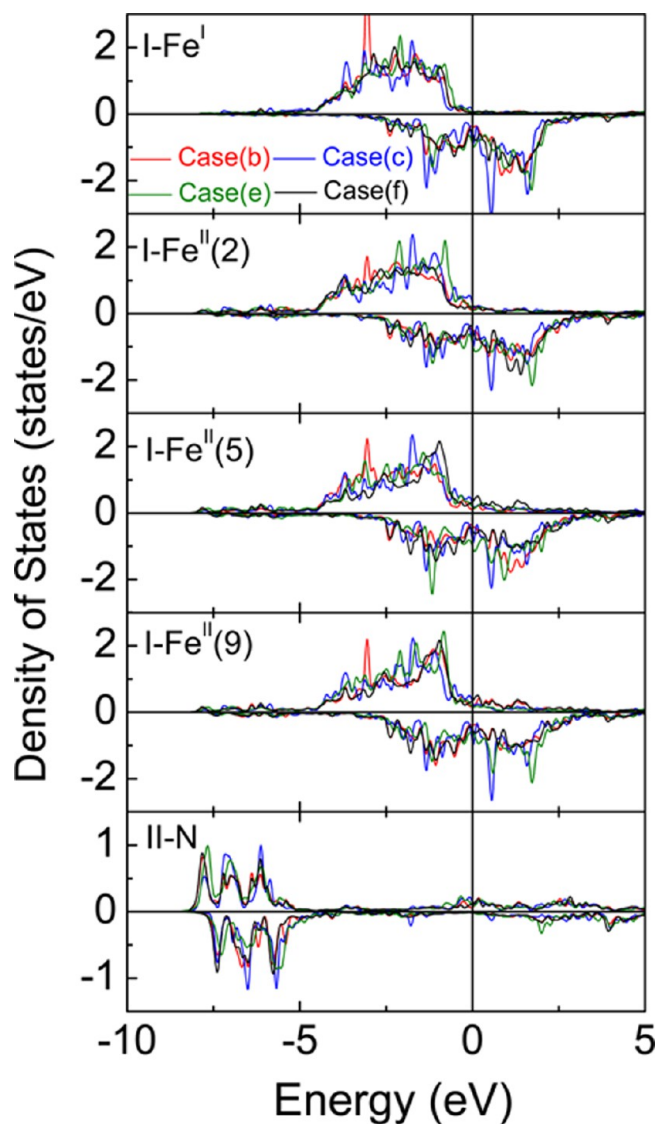


Figure 4. PDOS of Fe_4N in the fully relaxed $\text{Fe}_4\text{N}/\text{MoS}_2$ superlattice for model I, cases b, c, e, and f. The rows represent Fe^{I} and Fe^{II} in layer I and N in layer II. The Fermi energy is indicated by vertical lines.

except for layers I and VII in case e, see Table 3. The Fe^{II} atoms in layers I and VII hybridize with S in layers I, III and N in layers II, VI, respectively, which reduces the magnetic moment. The enhanced hybridization between Fe^{II} in layers III, V and N in layers II, IV, VI lowers the magnetic moment. Except for case c, the Fe^{II} magnetic moments in layers I and VII are slightly enhanced as compared to layers III and V because of stronger hybridization with N than S.

To highlight the influence of the stacking, we select cases b, c, e, and f as examples. Fe_4N layers I and VII are identical due to the periodicity of the supercell. The PDOSs in Figure 4 are obtained by projecting on the Fe^{I} , Fe^{II} , and N atoms at the interface, see Table 3 for corresponding magnetic moments. The smaller/larger Fe^{I} magnetic moments in case e/c are a consequence of the stronger/weaker hybridization with S just below E_{F} . Three Fe^{II} magnetic moments appear in layer I in cases b, e, and f, reflecting different hybridizations with S. Whereas the broken symmetry at the interface³⁸ promotes the Fe^{II} magnetic moment in layer I, the stronger hybridization between Fe^{II} and S weakens it. For case b, we find the strongest

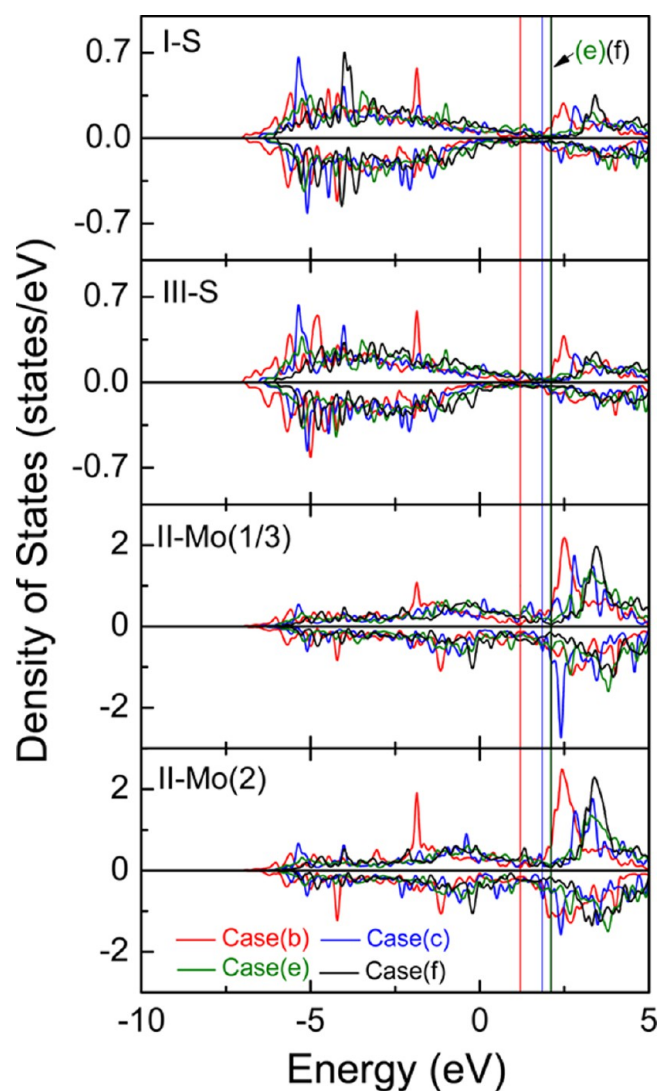


Figure 5. PDOS of MoS₂ in the fully relaxed Fe₄N/MoS₂ superlattice for model I, cases b, c, e, and f. The Fermi energy is indicated by vertical lines.

hybridization with S in layer I for Fe^{II}(9) followed by Fe^{II}(2) and Fe^{II}(5). For case c, the PDOS of the three Fe^{II} atoms in layer I is almost the same, reflecting a similar interaction with

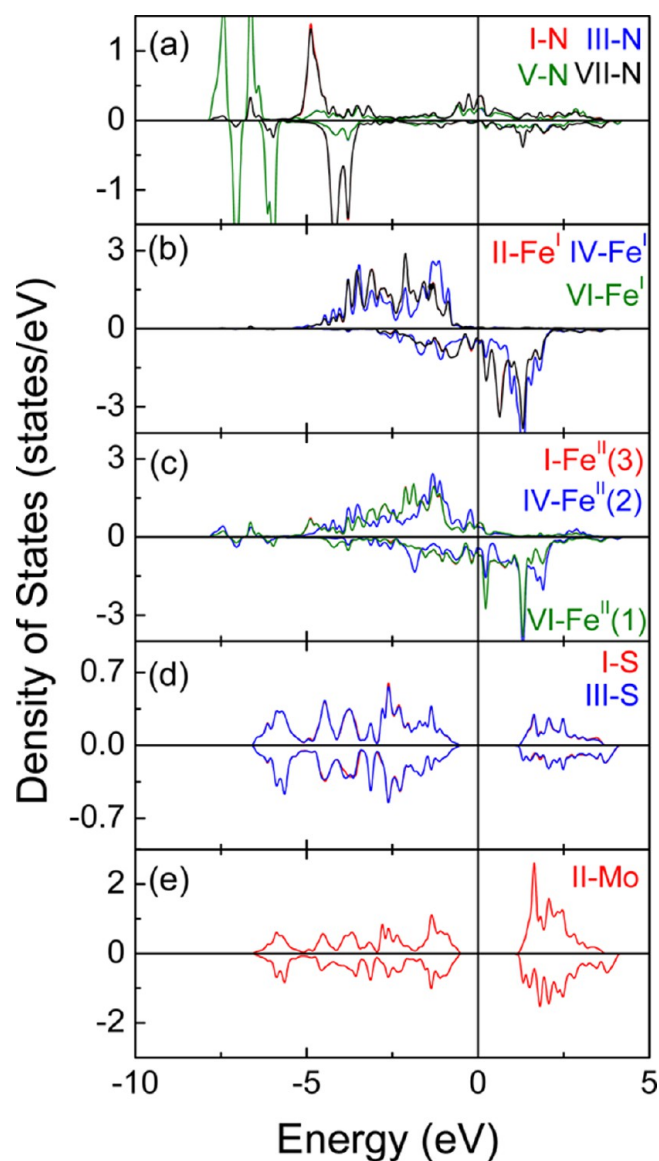


Figure 6. PDOS of the fully relaxed case g of the Fe₄N/MoS₂ superlattice: (a) N, (b) Fe^I, (c) Fe^{II}, (d) S, and (e) Mo. The Fermi energy is indicated by vertical lines.

Table 4. Average Fe^I and Fe^{II} Magnetic Moments \bar{m} (μ_B) for Each Fe₄N Layer in Model II; Magnetic Moments m (μ_B) of Fe^I, Fe^{II}, and N at the Interfaces As Well As Band Gaps E_g (eV) and p -Type Schottky Barrier Heights $\Phi_{B,p}$ (eV)

| moment/energy | atom | | bulk | g | h | i | j | k | l |
|---------------|------------------|----------|------|-------|-------|-------|-------|-------|-------|
| \bar{m} | Fe ^I | layer II | 2.95 | 2.98 | 2.99 | 2.03 | 2.98 | 2.98 | 2.99 |
| | | layer IV | 2.95 | 2.95 | 2.94 | 2.26 | 2.93 | 2.93 | 2.94 |
| | | layer VI | 2.95 | 2.98 | 2.99 | 2.05 | 2.98 | 2.98 | 2.99 |
| | Fe ^{II} | layer II | 2.35 | 2.17 | 2.12 | 2.27 | 2.05 | 2.05 | 2.12 |
| | | layer IV | 2.35 | 2.19 | 2.28 | 2.28 | 2.26 | 2.26 | 2.28 |
| | | layer VI | 2.35 | 2.17 | 2.12 | 2.28 | 2.02 | 2.02 | 2.12 |
| m | Fe ^I | (2) | 2.95 | 2.98 | 2.99 | 2.98 | 2.98 | 2.98 | 2.99 |
| | Fe ^{II} | (3) | 2.35 | 2.15 | 2.01 | 2.23 | 1.96 | 1.96 | 2.32 |
| | | (5) | 2.35 | 2.24 | 2.32 | 1.93 | 2.29 | 2.29 | 2.05 |
| | | (8) | 2.35 | 2.13 | 2.02 | 1.92 | 1.90 | 1.90 | 2.00 |
| | N | (1,2) | 0.03 | -0.01 | -0.02 | -0.03 | -0.03 | -0.03 | -0.03 |
| E_g | | | 1.89 | 1.87 | 1.87 | 1.86 | 1.87 | 1.86 | 1.86 |
| $\Phi_{B,p}$ | | | 0.54 | 0.54 | 0.54 | 0.59 | 0.59 | 0.59 | 0.54 |

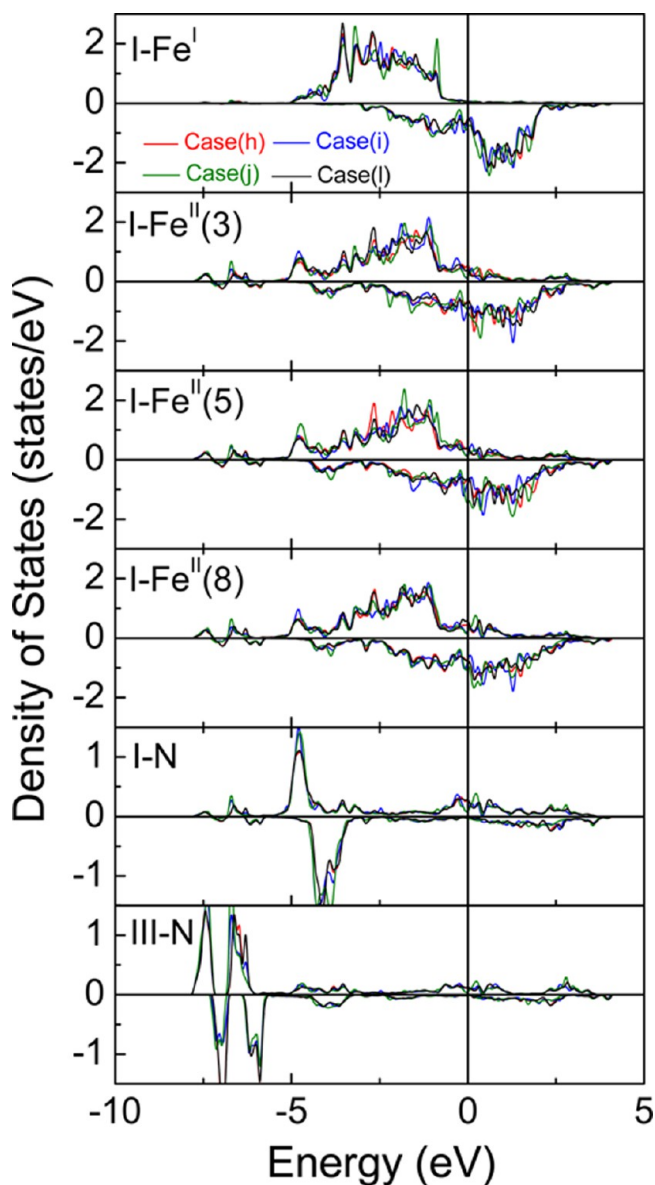


Figure 7. PDOS of Fe_4N in the fully relaxed $\text{Fe}_4\text{N}/\text{MoS}_2$ superlattice for model II, cases h, i, j, and l. The rows represent Fe^{I} and Fe^{II} in layer I and N in layers I and III, respectively. The Fermi energy is indicated by vertical lines.

MoS_2 , whereas in case e S hybridizes more with $\text{Fe}^{\text{II}}(5)$ than with $\text{Fe}^{\text{II}}(9)$ and even less with $\text{Fe}^{\text{II}}(2)$, as shown in Figures 4 and 5. This order is the same in case f, in which the hybridization of $\text{Fe}^{\text{II}}(5)/\text{Fe}^{\text{II}}(9)$ with S is smaller/larger than in all other cases.

Figure 5 gives Mo and S PDOSs for model I. It is clear that the electronic states of S are similar in layers I and III, where the interaction with Fe_4N has dramatic effects near E_F . Strong orbital overlap between Fe^{I} , Fe^{II} , S, and Mo induces metallicity in MoS_2 . The bonds between S (layer I) and $\text{Fe}^{\text{I}}/\text{Fe}^{\text{II}}$ (layer I) weaken the Mo–S bonds, which leads to the spin polarization with a small Mo magnetic moment, see Table 3. Clearly, the magnetic moments depend on the position, with the highest value of $0.22 \mu_B$ in case c, where Mo sits directly on top of Fe^{I} . The total magnetic moments of MoS_2 are 0.14, -0.02 , 0.33, 0.10, -0.11 , and $-0.02 \mu_B$ in cases a–f. Denoting the total DOS in the spin-up and spin-down channels by N_{\uparrow} and N_{\downarrow} , the spin

polarization $P = [N_{\uparrow}(E_F) - N_{\downarrow}(E_F)]/[N_{\uparrow}(E_F) + N_{\downarrow}(E_F)]$ can be derived from Figure 5 as $P = -40, -52, -52$, and -44% . The fact that the sign is the same as in Fe_4N demonstrates that the polarization is not inverted.

The charge transfer at the interfaces is quantified by charge density difference calculations (see Figure 1), showing that charge is lost from the Fe and S atoms, while it accumulates in the Fe–S bond region, reflecting covalent bonding across the interface. It is worth noting that some extra charge accumulates around Mo. We find more charge in the covalent $\text{Fe}^{\text{I}}\text{–S}$ bonds for case e as compared to all other cases, consistent with the shortest bond length (Table 1) and the smallest Fe^{I} magnetic moment [Table 3]. For cases a and c, the charge distributions between Fe^{II} and S in layer I are almost identical (same Fe^{II} magnetic moment), whereas in cases b, e, and f the amount of charge accumulated in the $\text{Fe}^{\text{II}}\text{–S}$ bonds corresponds perfectly to our previous analysis of Fe^{II} magnetic moments. The stronger the covalent bond is, the smaller the Fe^{II} magnetic moment is.

For model II the Fe_4N magnetic moments are subject to reductions of 5.14, 5.42, 7.40, 7.42, 7.42, and 5.36% as compared to bulk Fe_4N for cases g–l, respectively. The obtained average Fe^{I} and Fe^{II} magnetic moments are presented in Table 4. The electronic properties of MoS_2 are similar for the different stackings due to similar interactions with Fe_4N . We select case g as an example to elaborate on the similarities in Figure 6. Hybridization appears between N in layer I and S in layer I, whereas Fe has hardly any hybridization with S because of the large distance. As compared to model I, the Fe^{I} magnetic moment in layers II and VI is larger than bulk value. The Fe^{II} magnetic moment, on the other hand, is smaller, where the reduced values in layers II and VI can be ascribed to the enhanced hybridization with N in layers I, III, V, and VII. The weak interfacial bonds induce no significant Mo and S magnetic moments ($0.001 \mu_B$).

As cases k and j behave similarly, we use cases h, i, j and l to study the stacking effect. The magnetic moments of Fe^{I} , Fe^{II} , and N at the interfaces in model II are shown in Table 4. The three values of Fe^{II} magnetic moments in layer II reflect different hybridizations with N in layers I and III. From Figure 7, there is less hybridization between $\text{Fe}^{\text{II}}(5)$ and N in layers I and III (larger magnetic moment) in case h, as compared to $\text{Fe}^{\text{II}}(3)$ and $\text{Fe}^{\text{II}}(8)$. In case i $\text{Fe}^{\text{II}}(3)$ shows the smallest hybridization with N in layer III, while all Fe^{II} atoms interact similarly with N atoms in layer I, as shown in Figure 7. Similarly, in case j, the enhanced $\text{Fe}^{\text{II}}(5)$ magnetic moment is due to reduced hybridization with N in layer III. The interaction with layer I is almost the same for all the Fe^{II} atoms. Finally, case l resembles case i.

Figure 8 gives band structures for pristine monolayer MoS_2 and the superlattices, demonstrating the changes in the electronic states of MoS_2 upon interface formation. In Figure 8a, the original K point of the 1×1 unit cell of monolayer MoS_2 is folded to the Γ point of the $\sqrt{3} \times \sqrt{3}$ cell. In case c the MoS_2 bands strongly hybridize with Fe_4N , inducing a metallic character of monolayer MoS_2 , as shown in Figure 8b, c. For case g, although MoS_2 hybridizes with Fe_4N to a certain extent, the gross characters of the MoS_2 -related bands are almost identical to the free-standing monolayer, see Figure 8d, e, indicating a weak interaction. Furthermore, in case g E_F is located 0.56 eV above the VB maximum and 1.37 eV below the CB minimum of MoS_2 , indicating p-doping. In model II, interestingly, the semiconducting nature of monolayer MoS_2 is

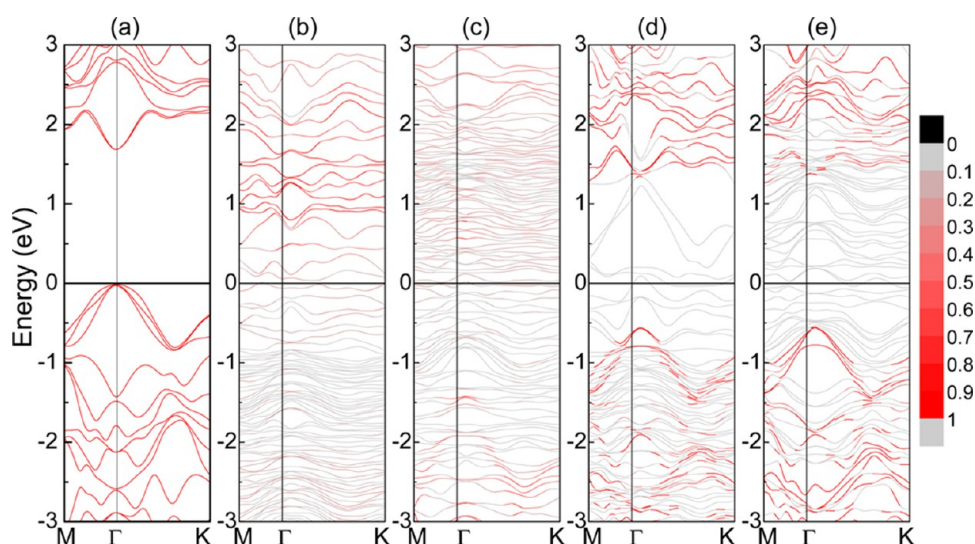


Figure 8. (a) Band structure of a $\sqrt{3} \times \sqrt{3}$ superlattice of monolayer MoS₂. (b, d) Majority and (c, e) minority spin band structure for cases (b, c) and (d, e) g. Red color indicates MoS₂ and gray Fe₄N. The Fermi energy is indicated by vertical lines.

preserved. The band gap increases as compared to model I, see Table 4.

Because E_F lies in the MoS₂ band gap, a Schottky barrier is formed at the interface for model II, which is of crucial importance for the device performance. The p-type Schottky barrier heights ($\Phi_{B,p}$) are the energy differences between the VB maximum of MoS₂ and E_F of the superlattice. We obtain values of 0.54, 0.54, 0.59, 0.59, 0.59, and 0.54 eV for cases g–l, respectively (Table 4). An ideal Schottky junction fulfills $\Phi_{B,p} = E_g - (\Phi_M - \chi_s)$, where E_g and χ_s are the band gap and electron affinity of the semiconductor and Φ_M is the work function of the metal.³⁹ From the vacuum potential and E_F , we obtain $\Phi_M = 5.38$ eV for the N-terminated surface of Fe₄N(111). We derive $\chi_s = 4.27$ eV, which is identical to a reported value.¹⁷ The resulting $\Phi_{B,p} = 0.57$ eV deviates only slightly from the value derived from the band structure, reflecting weak Fermi level pinning and proximity to the Schottky limit. In general, Fermi level pinning is governed by metal induced gap states,⁴⁰ surface states,⁴¹ and dipole effects.⁴² Figure 1 shows significant charge accumulation around the interfacial N atoms and charge depletion at S atoms. Therefore, the p-type doping of MoS₂ on the N-terminated surface of Fe₄N(111) is due to electron transfer from MoS₂ to Fe₄N. Even in cases h and j, where the charge is more localized, there is a transfer to N.

CONCLUSION

The main findings of our first-principles calculations on Fe₄N(111)/MoS₂($\sqrt{3} \times \sqrt{3}$) superlattices are summarized as follows: For model I: (1) strong Fe–S chemical bonds are formed; (2) the Fe^I and Fe^{II} magnetic moments in the Fe₄N region are reduced; (3) spin-polarization is induced in MoS₂, with the largest magnetic moments of 0.33 μ_B for Mo located on top of the interfacial Fe^I atoms. For model (II): (1) interfacial N–S bonding is relatively weak; (2) the Fe^{II} magnetic moments are reduced; (3) the semiconducting nature of MoS₂ is preserved and the band gap increases. Importantly, we find p-type doping of MoS₂ with Schottky barrier heights in the range of 0.5–0.6 eV for different stackings. Therefore, Fe-terminated Fe₄N in contact with monolayer MoS₂ has potential in spintronics devices, whereas N-termination is interesting for high-performance p-type field effect transistors.

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Author Contributions

N.F. and W.M. designed the outline of the manuscript and wrote the main manuscript text. Y.C., Z.G., U.S., and H.B. contributed detailed discussions and revisions. All authors reviewed the manuscript.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

W.B.M. was supported by the National Natural Foundation of China (51172126), Key Project of Natural Foundation of Tianjin City (12JCZDJC27100), Program for New Century Excellent Talents in University (NCET-13-0409), and Scientific Research Foundation for the Returned Overseas Chinese Scholars, State Education Ministry of China. Y.C.C. and U.S. were supported by a CRG grant of KAUST.

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